



**ED-307**

M.Sc. 1st Semester  
Examination, March-April 2021

**CHEMISTRY**

Paper - III

Quantum Chemistry : Thermodynamics and  
Chemical Dynamics - I

*Time* : Three Hours]      [*Maximum Marks* : 80

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**Note** : Answer **all** questions. The figures in the right-hand margin indicate marks.

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**Unit-I**

1. (a) Find the inverse of the matrix 5

$$A = \begin{bmatrix} 3 & -2 & -1 \\ -4 & 1 & -1 \\ 2 & 0 & 1 \end{bmatrix}$$

(b) Outline the variation method used for obtaining approximate value of ground state energy system. 5

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*(Turn Over)*

( 2 )

- (c) What are the various methods for obtaining approximate solution to the wave function equation? Discuss the perturbation method and application of first order perturbation theory of the atom. 10

**OR**

- (a) Find the inverse of the matrix 5

$$\begin{bmatrix} 5 & -2 & 3 \\ 4 & -1 & -5 \\ 6 & 7 & 9 \end{bmatrix}$$

- (b) Explain angular momentum operator. Work out measurable value of the angular momentum of the particle. 5
- (c) Describe eigen value and matrix element of angular momentum operator. 10

**Unit-II**

2. (a) What is partition function? Discuss rotational partition function. 5
- (b) Describe Maxwell's thermodynamical relation and discuss its application in proving  $C_p - C_v = R$ . 15

**OR**

- (a) Explain partial molar volume and partial molar heat content. 5

( 3 )

- (b) What is the most probable distribution ? Explain Maxwell-Boltzmann distribution law of energy partition. Compare it with Bose-Einstein statics. 15

**Unit-III**

3. (a) Write notes on the following : 10  
(i) Activity coefficient  
(ii) Electro-catalysis  
(b) Explain electrical double layer. Discuss Gouy-Chapman electrical double layer. 10

**OR**

- (a) Describe the following : 10  
(i) Ionic strength  
(ii) Over potential  
(b) Discuss Debye-Huckel theory for activity coefficient of electrolytic solution. 10

**Unit-IV**

4. (a) What are the fast reactions ? Describe flash photolysis method of studying fast reaction. 10  
(b) Discuss the following : 10  
(i) Activated complex theory  
(ii) Oscillatory reactions

**OR**

( 4 )

- (a) Describe Lindemann's theory of unimolecular reactions. 10
- (b) Discuss the following : 10
- (i) Secondary salt effect
- (ii) Rate expression for the photochemical reaction of  $H_2$  and  $Br_2$
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